



Understanding Your Fertilizer Types and Composition

Volume 13, Issue 6

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June 2020

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Fertilization in forage systems can account for the large cost of maintaining pastures or increasing hay production. It is important to remember that all soils are not created equal and your forage productivity will depend on the soil's capacity and chemical properties (especially pH) to supply nutrients. Forage nutrients requirements also depend on how much nutrients are being removed from the soil over time and how often those nutrients are replenished through a fertility management program. Keep in mind that different forages have different nutrient contents. Nutrient removal, especially in hay production, can be estimated by using the average nutrient content of the forage species and the yield produced per acre. Keep in mind that nutriment removal will increase as yield production increases. This means that over time the nutrient applications will be the greatest in forage and soils with the highest yields. That is also a reason why a soil test recommendation has a three-year plan.



Figure 1. Fertilizer bag label information.

Fertilizer use is an important factor for producing profitable forages and for maintaining soil nutrient balance over

the years. The first step to developing a fertility program for your forage system must always start with a soil sample to determine nutrient needs. Most producers commonly make fertilizer applications without any soil nutrient information, which means the possibility of over-fertilizing or under-fertilizing. Either approach will have an economic impact on your farm. Nitrogen, phosphorus, and potassium are called primary macronutrients because they are needed in large quantities to support forage production at profitable economic levels and to replace nutrients removed with each hay harvest.

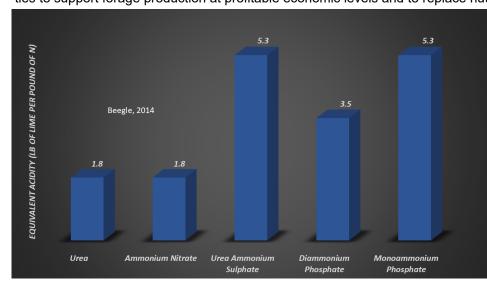


Figure 2. Equivalent acidity for different fertilizer products.

Selecting a fertilizer type is also important in forage production.

Fertilizer Functions – Most fertilizer suppliers will have different fertilizer types available depending on the source, region, and environmental conditions for utilizing those fertilizer types. By understanding the basics behind fertilizer labels, you can choose the right fertilizer type(s) for your forage system based on soil nutrient availability, forage crop, and expected yields. Nitrogen and potassium are usually required in larger quantities on forage systems than phosphorus because they can be taken up in larger quantities by the forage crop or lost through different processes. Phosphorus tends to be very stable in the soil and it

is required in lower quantities than N and K.

Nitrogen is needed for biomass production, increase leaf growth, and increase crude protein. Phosphorus (P) is essen-

Table 1. Readily available fertilizers for forage production in Mississippi.			
Fertilizer Source	Fertilizer Type	Fertilizer Grade	Characteristics
	Ammonium Nitrate	34-0-0	Dry granular material Contains 680 lbs N/ton Readily available N 100% water soluble Low volatilization
	Urea	46-0-0	Dry granular material Contains 980 lbs N/ton Readily available N 100% water soluble High volatilization in summer Apply close to a rainfall event to move into the soil and reduce volatilization losses
z	Urea Ammonium Sulfate	33-0-0	Dry granular material 50/50 blend of urea and ammonium sulfate Contains 660 lbs N/ton Readily available N 100% water soluble Medium volatilization - Urea portion subject to volatilization losses
	Urea Ammonium Nitrate Solution	32-0-0	Clear liquid, caustic material Contains 640 lbs N/ton Caustic solution burns green tissue to which it is applied Medium volatilization - Urea portion subject to volatilization losses
	Diammonium Phosphate (DAP)	18-46-0	Dry granular material Readily available N and P 100% water soluble
۵	Monoammonium Phosphate (MAP)	11-52-0	Dry granular material Readily available N and P 100% water soluble
	Triple Super Phosphate	0-46-0	Dry granular material Readily available P 100% water soluble
¥	Potassium Chloride or Muriate of Potash	0-0-60	Dry granular material Readily available K 100% water soluble

tial for seed germination, strong root growth, and seed development. Potassium (K) enhances overall growth while helping to regulate root and top growth and increasing plant protection to pest resistance and drought and cold tolerance.

Calculating Your Fertilizer Nutrient Content - All fertilizer types must provide uniform information that will help forage producers to compare products. Every bag of fertilizer contains three names or abbreviations in the following order Nitrogen (N), Phosphorus (P₂O₅), and potassium (K₂O) (Fig. 1). They represent the oxide form for elemental phosphorus (P), and potassium (K). These can be converted to elemental $P (P = P_2O_5 \times 0.44 \text{ or } P_2O_5 = P \times 2.3) \text{ or}$ elemental K (K = $K_2O \times 0.83$ or $K_2O = K \times 10^{-2}$ 1.2). These letters are also associated with three numbers that represent an N-P-K ratio and the proportion of each nutrient in the same order (N-P-K). Fertilizer ratio is the ratio of weight percent of N-P₂O₅-K₂O and is determined by dividing the three numbers by the smallest of the three. Fertilizer manufacturers usually produce different fertilizer grades depending on the market (turf vs. forages). For example, a bag of fertilizer with a 34-46-60 means that a 100-bag contains 34% nitrogen, 46% phosphorus, and 60% potassium by weight. In order words, that 100-bag contains 34 lbs of nitrogen $(0.34 \times 100 = 34)$, 46 lbs of phosphorus $(0.46 \times 100 = 46)$, and 60 lbs of potassium $(0.60 \times 100 = 60)$. Usually, this fertilizer grades can be formed by using only source (34-0-0, 46-0-0, 0-46-0, 0-0 -60) or by mixing two or more nutrient sources to form a blend (0-20-20, 18-46-0, 33-0-0S). Fertilizer blends are usually called "mixed fertilizers" and they contain particles of more than one color.

Selecting a Fertilizer Grade for Forage Production – The best way to select a fertilizer grade is to have your soil tested. Soil test all fields on a 2 (hay) to 3-year

(pasture) interval. Soil samples should be collected at the same time during the year to avoid seasonal variation in soil pH levels and nutrient availability. Phosphorus and potassium fertilizers should be applied according to soil test recommendations (Table 1). These recommendations provide for building soil fertility levels when soil samples show that the nutrient levels are low and maintaining high levels when the high fertility has been achieved. In general, nitrogen fertilizer application to pastures or hay should be split to increase efficiency. It is recommended that to apply 60 units of N per acre in split applications (after green-up) and mid-summer after the second grazing cycle. For hay production, nitrogen applications of 50 units per acre per cut of hay should be sufficient when hay is cut in a 28 to 35-day cycle. However, N applications in the summer should be carefully timed with plenty of moisture along with choosing the correct nitrogen source to reduce losses. Producers cannot overcome a lack of rainfall with increased fertilization. Table 1 provides some information about the types of nitrogen sources that are available for use in forage production in Mississisppi.

Sometimes producers are concerned with the production of acidity by the reaction of fertilizer materials, especially from N fertilizers. The acidity of a fertilizer is usually expressed as the amount of pure lime that would be required to neutralize the acidity produced by the reaction of the fertilizer. Figure 2 indicates that highest amount of acidity is not necessarily created by nitrogen source, but amount and number of applications. Although phosphate-based fertilizer has higher acidity, they are applied less often and smaller quantities than N based fertilizers.

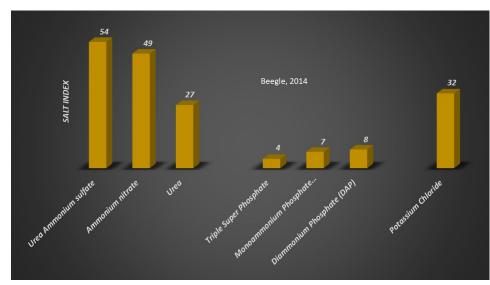


Figure 3. Salt index for different fertilizer products. The salt index assumes that primary weights of the initial nutrient are being compared.

Another property to consider when selecting a fertilizer source is the amount of soluble salt in the fertilizer. High concentration of soluble salts can cause plant injury and prevent seed germination. Application of fertilizers at the recommended rates usually do not have high levels of soluble salts to cause plant injury; however, when fertilizer are mixed with seed for planting, soluble salt levels that are high enough can cause seed embryo damage and reduce germination. The salt index of several fertilizer types is shown in Figure 3.

Finally, the selection of a nitrogen fertilizer source should be based on the cost per pound of N and the potential for volatilization losses. For example, using local fertilizer prices for urea (\$382/ton, \$0.61/lb N), ammonium ni-

trate (\$350/ton, \$0.51/lb N), urea ammonium sulfate (\$415/ton, \$0.63/lb N), and 32% UAN solution (\$294/ton). This means that the cost per acre for applying 50 lbs N/ac will be \$30.50 for urea, \$25.50 for ammonium nitrate, and \$31.50 for urea ammonium sulfate, and \$23.0 for UAN solution. In many situations, the cheapest fertilizer cost or the fertilizer with a higher N grade might not be the best option. It is also important to consider weather conditions (moisture, heat, and humidity) that can impact the potential for volatilization losses. For example, urea can have volatilization losses as high as 20 to 30% in warm temperatures with no rainfall at application time or soon after. A reason why urea is more suited for applications to cool-season grasses during the winter, spring, and late fall.

Summary – Developing a well-managed and efficient forage fertilization program that can increase profitability routinely depends on soil testing as well as maintain records of the applied fertilizers and nutrient removal. The amount of nitrogen to be applied per unit should be based on expected yields and the rainfall pattern during the season along with split applications to increase utilization and efficiency. It is also important to select the appropriate fertilizer source(s) to supply the need nutrient at a lower cost.

Upcoming Events

August 4, 2020—Herd Health and Forage Field Day, Raymond, MS For registration contact Melissa Washburn at 601-857-3334 or mcwashburn@hindscc.edu

For upcoming forage related events visit: http://forages.pss.msstate.edu/events.html

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