

# Soil pH and Tree Species Suitability in Mississippi



The suitability of the species of tree you wish to grow depends primarily on the soil characteristics of your site. Among the many soil properties, soil pH is one of the most limiting. Soil pH provides a good indication of the chemical status of the soil and can help determine potential plant growth. This publication is designed to help landowners and foresters gain a better understanding of soil pH and its effects on species–site relationships in Mississippi.

## What Is Soil pH?

Soil pH is a measure of how acidic or basic a soil is. The pH scale runs from 0 to 14 and is a measure of the hydrogen ion concentration of a solution. Acidic pH values are those less than 7.0, neutral pH values are 7.0, and basic pH values are those greater than 7.0. Normally, pines grow best on acidic soils, and hardwoods prefer slightly acidic to neutral soils. However, most tree species can grow well over a broad range of pH values.

## Why Is Soil pH Important in Forestry?

Soil pH influences nutrient uptake and tree growth. Many soil nutrients change form during reactions in the soil that are largely controlled by soil pH. Trees may or may not be able to use these changed nutrients. Soils with a pH of 6.5–7.0 generally provide the best growing conditions. In this pH range, most nutrients are readily available. The vast majority of commercially important tree species can live in a broad soil pH range if the proper balance of required nutrients is maintained.

Soil pH values at the extremes (<4.0 and >8.5) can create toxic nutrients and make necessary nutrients unavailable to plants. At lower pH levels (<5.0), aluminum, iron, and manganese are readily available for plant uptake. At higher pH levels (>7.5), calcium and potassium are overabundant. In these situations, many plants will take up too many of these nutrients but not enough of others. This imbalance causes toxic conditions. **Table 1** shows a range of soil pH values and nutrient availability within that range.

**Table 1. Soil pH effects on soil nutrient availability.**

Columns with dark fill show the greatest nutrient availability, hatched spaces represent moderate nutrient availability, and blank spaces represent limited availability for that pH level.\*

	pH	4.0	4.5	5.0	5.5	6.0	6.5	7.0	7.5	8.0	8.5	9.0	9.5	10.0
Nitrogen (N)			hatched	hatched	hatched	dark	dark	dark	dark	dark	hatched	hatched	hatched	
Phosphorus (P)					hatched	hatched	dark	dark	dark	hatched	hatched	dark	dark	dark
Potassium (K)					dark	dark	dark	dark	dark	dark	dark	dark	dark	dark
Calcium & Magnesium (Ca & Mg)		hatched	hatched	hatched	dark	dark	dark	dark	dark	dark	dark	dark	dark	dark
Sulfur (S)					dark	dark	dark	dark	dark	dark	dark	dark	dark	dark
Boron (B)			hatched	dark	dark	dark	dark	dark	hatched	hatched	hatched	dark	dark	dark
Copper & Zinc (Cu & Zn)				dark	dark	dark	dark	dark	dark	dark	dark	dark	dark	dark
Molybdenum (Mb)					hatched	dark	dark	dark	dark	dark	dark	dark	dark	dark
Iron & Manganese (Fe & Mn)		dark	dark	dark	dark	dark	dark	dark	hatched	hatched	dark	dark	dark	dark
Aluminum (Al)		dark	dark	hatched	dark	dark	dark	dark	dark	dark	dark	dark	dark	dark

\*Adapted from Plaster, E.J. 1996. Soil Science and Management. Third Edition. Delmar Publishers, Albany, New York. 402 pp.

## How Is Soil pH Determined?

You can determine the pH of your property's soil by two methods. The most accurate way is to collect a sample of soil and have it analyzed. The MSU Extension Service offers this service for a modest fee. For detailed information on how to properly collect soil samples and have them tested, please refer to Information Sheet 1294 *Soil Testing for the Homeowner*. You can also contact your county Extension agent for procedural details on soil testing. The second method of determining your property's pH involves looking in your county soil survey manual. The Natural Resources Conservation Service (NRCS) has conducted county soil surveys for many years. Your local NRCS office can provide you with a copy of the manual and help you use it.

Fortunately, many trees grow successfully and survive over a wide range of soil pH levels. However, some species grow better at a particular soil pH level.

Table 2 lists some tree species common to Mississippi and the soil pH range at which they have the best growth potential. The pH values provided below are intended to serve as a guide only. These species will often grow on soils outside the pH range listed in the table.

## Summary

Soil pH is a measure of the acidity or alkalinity of soil. Levels of soil pH can be used as a general guide for determining what species will grow on a given site and are typically indicative of available nutrient levels. Use this publication as a guide to determine site suitability regarding soil pH. It only considers soil pH as a factor affecting tree survival and growth. Other soil conditions also have major influences on the suitability of a site for various tree species. For more information on other soil conditions that can limit growth, please see Publication 2004 *Bottomland Hardwood Management Species-Site Relationships*, or contact your area Extension forestry specialist.

**Table 2. Preferred soil pH ranges for Mississippi tree species.\***

Common Name	Scientific Name	pH Range	Common Name	Scientific Name	pH Range
Ash, Green	<i>Fraxinus pennsylvanica</i>	3.6–7.5	Oak, Shumard	<i>Q. shumardii</i>	4.4–6.2
Baldcypress	<i>Taxodium distichum</i>	4.6–7.5	Oak, Southern Red	<i>Q. falcata</i>	5.0–7.0
American Beech	<i>Fagus grandifolia</i>	5.0–7.5	Oak, Water	<i>Q. nigra</i>	3.6–6.3
River Birch	<i>Betula nigra</i>	4.5–6.0	Oak, White	<i>Q. alba</i>	4.5–6.2
Blackgum	<i>Nyssa sylvatica</i>	4.6–7.0	Oak, Willow	<i>Q. phellos</i>	3.6–6.3
Cottonwood	<i>Populus deltoides</i>	3.6–7.5	Paulownia	<i>Paulownia tomentosa</i>	6.0–8.0
Dogwood	<i>Cornus spp.</i>	5.0–8.0	Pecan	<i>Carya illinoensis</i>	4.8–7.5
Hackberry	<i>Celtis occidentalis</i>	5.0–7.5	Persimmon	<i>Diospyros virginiana</i>	4.4–7.0
Hickory	<i>Carya spp.</i>	4.5–7.5	Pine, Loblolly	<i>Pinus taeda</i>	4.5–7.0
Magnolia, Southern	<i>Magnolia grandiflora</i>	5.0–6.0	Pine, Longleaf	<i>P. palustris</i>	4.5–7.0
Maple, Red	<i>Acer rubrum</i>	4.4–7.5	Pine, Shortleaf	<i>P. echinata</i>	4.5–7.0
Oak, Cherrybark	<i>Quercus pagodafolia</i>	4.5–6.2	Pine, Slash	<i>P. elliotii</i>	4.5–7.0
Oak, Live	<i>Q. virginiana</i>	6.0–7.5	Redcedar, Eastern	<i>Juniperus virginiana</i>	6.0–7.5
Oak, Northern Red	<i>Q. rubrum</i>	4.5–6.0	Sweetgum	<i>Liquidambar styraciflua</i>	3.6–7.5
Oak, Nuttall	<i>Q. nuttallii</i>	3.6–6.8	Sycamore	<i>Platanus occidentalis</i>	4.4–7.5
Oak, Post	<i>Q. stellata</i>	5.0–7.5	Walnut, Black	<i>Juglans nigra</i>	5.0–7.5

\*Adapted from Williston, H.L., and R. LaFayette. 1978. Species Suitability and pH of Soils in Southern Forests. USDA Forest Service. Southeastern Area, State and Private Forestry. Forest Management Bulletin. 4 pp.

Publication 2311 (POD-03-16)

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Extension Service of Mississippi State University, cooperating with U.S. Department of Agriculture. Published in furtherance of Acts of Congress, May 8 and June 30, 1914. GARY B. JACKSON, Director